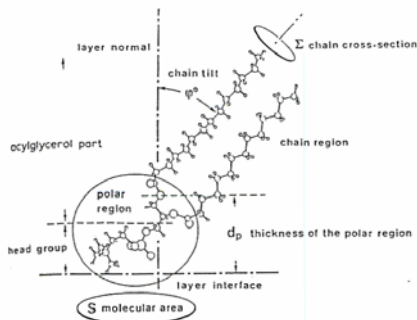


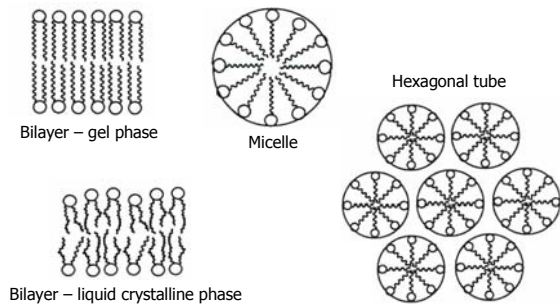
Lipid and DNA Modeling

General Lipid Structure



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Lipid Aggregates (Phases)



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Greatest Challenge

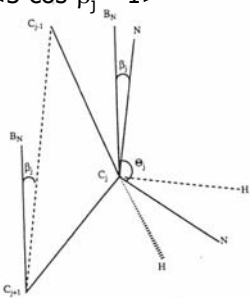
- Setting up an appropriate starting system
 - Conformation of lipid chains
 - Surface area/lipid head group
 - Solvation

Acyl Chain Disorder

- Liquid crystalline phase lipids (most relevant for studying biological membranes) have increased mobility relative to gel phase lipids
- Increased mobility gives rise to 'gauche defects' which are found in increasing concentration toward the center of the bilayer
- The acyl chain disorder can be represented by statistically averaged molecular order parameters

Molecular Order Parameters

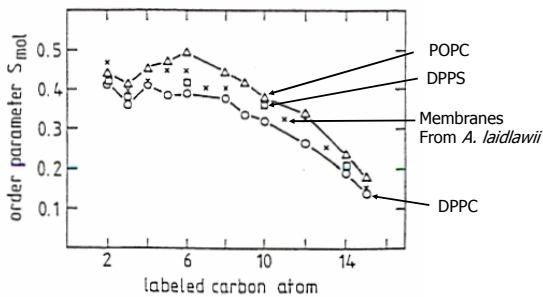
- $S_j^{\text{mol}} = 0.5 \langle 3 \cos^2 \beta_j - 1 \rangle$



Experimental Order Parameters

- ^2H NMR with deuterium labeled lipids
 - Series of measurements made on lipids with deuterium at different positions down the chain
 - Angle measured is between magnetic field and carbon-deuterium bond
 - NMR order parameter profile must be multiplied by -0.5 for comparison to previously defined profile

Example Experimental Order Profile




Lipid Surface Areas

- Examples
 - Dilauroylphosphatidylethanolamine (DLPE)
 - Ammonium head group
 - $39-51 \text{ \AA}^2$
 - Dimyristoylphosphatidylcholine (DMPC)
 - Tetraalkylammonium head group
 - $60-70 \text{ \AA}^2$



Hydration

- Most lipid headgroups are hydrated with water and may require counterions
- Bilayer surfaces are subject to hydration pressure when they are brought into close proximity
- These repulsive forces may require substantial layers of water if periodic boundary conditions are used



DNA Structure – Class Exercise

- Download a segment of double-stranded DNA from the protein databank
- Examine the structure for the following features:
 - Charged groups
 - Hydrogen bonding interactions
 - Overall morphology (shape)



Greatest Challenge

- Electrostatic Treatment and Counterions
 - Polyanionic DNA chain is surrounded by a cloud of ions that compensate for the concentration of anionic groups
 - This ion cloud is referred to as the ion atmosphere
 - The ions are mobile as they are not covalently attached to any particular phosphate group

Manning Theory

- $\xi = q^2 / (\epsilon k T b)$
 - q = charge on the counterion
 - ϵ = solvent dielectric
 - k = Boltzmann constant
 - T = temperature
 - b = distance between backbone phosphates along axis
- Net charge on phosphate = $1 / (N \xi)$ where N is the valency of the counterion
 - -0.24 with Na^+ counterions
 - -0.12 with Mg^{2+} counterions

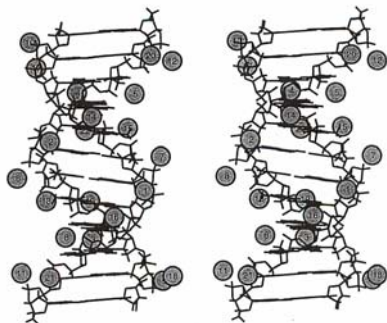
Applying Manning Theory

- DNA simulations lacking explicit counterions utilize Manning Theory to assign charges
 - Usually to phosphorous and attached oxygens
 - Either scaled by a factor of 0.24-0.34
 - Or assigned to sum to -0.34
 - Sometimes all charges in DNA scaled by 0.25

Explicit Counterions

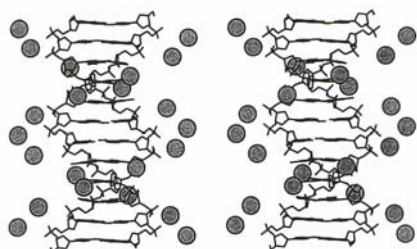
- Placement – First method
 - Solvate DNA
 - Compute electrostatic potential (EP) on each water
 - Replace those with highest negative EP with counterions
- Placement – Second method
 - Calculate electrostatic potential around DNA
 - Place counterion at grid point with highest negative EP
 - Repeat
- Placement – Third method
 - Place ions 4.5-6.0 Å from P bisecting the O-P-O angle

Placement Example (Method 2)



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Placement Example (Method 3)



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Counterion Equilibration

- Regardless of placement method, counterion positions need sufficient equilibration to find optimal positions
- Often such equilibration is performed while holding the DNA fixed and just allowing the water and counterions to relax

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Subsequent Modeling (DNA and Lipid)

- Once the initial challenges are met, subsequent modeling of these systems can be done with methods we've already discussed



Further (optional) Reading

- Reviews in Computational Chemistry, volume 11, chapter 6 (DNA counterion treatment)
- Reviews in Computational Chemistry, volume 5, chapter 5 (Lipid simulations)
