



# Breaking strength-ductility trade-off in laser-powder bed fused Fe–Cr–Ni–Al maraging stainless steel: Controlled precipitation and preserved dislocations

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## ABSTRACT

In the current study, cylindrical rods of Fe–Cr–Ni–Al maraging steel (with the brand name CX) are fabricated using the laser-powder bed fusion (L-PBF) technique. The material is analyzed using the differential scanning calorimetry (DSC) technique under continuous heating at different heating rates. The DSC results are employed to identify the heat flow peaks associated with the precipitation and austenite reversion phase transformations. The peaks are then analyzed and processed to determine the onset, peak, and end temperatures associated with each phase transformation. These results are used to calculate the degree of phase transformation under non-isothermal conditions and to evaluate the activation energy of transformation through the Kissinger method. In the next step, the modified Johnson-Mehl-Avrami-Kolmogorov (JMAK) model is employed to model the kinetics of phase transformations during non-isothermal heating. The validated model is then used to predict the kinetics of precipitation and austenite reversion phase transformations under isothermal aging heat treatments. The isothermal modeling results are employed to design a direct aging heat treatment to promote the evolution of nanometric  $\beta$ -NiAl precipitates and meanwhile preserve the pre-existing dislocation networks. Such a heat treatment results in the enhancement of both yield and tensile strengths from 929 MPa and 1032 MPa in the as-built condition to 1659 MPa and 1738 MPa after heat treatment. Meanwhile, ductility slightly changes from 15.3% in the as-built condition to 14.7% after heat treatment. Such an achievement in breaking the strength-ductility trade-off is due to a controlled heat treatment process that preserves the dislocation networks.

## 1. Introduction

Precipitation-hardenable (PH) stainless steels are of particular interest for applications under severe service conditions such as load-bearing at high temperatures, and corrosive environments. Therefore, they possess vast applications in advanced industries such as aerospace, marine, and nuclear [1,2]. PH stainless steels have significantly higher strength than ferritic or austenitic stainless steels [3]. Meanwhile, these steels possess good ductility and toughness [4]. Different classes of PH stainless steels have been developed by changing the chemistry and the resulting precipitates and properties. Some examples include PH 13-8 Mo, PH 17-4, and PH 15-5. Among these alloys, PH 13-8 Mo possesses superior mechanical properties compared to other types [5].

In the case of the existence of a martensitic microstructure, similar to the microstructure of PH 13-8 Mo, the steel is called maraging stainless steel since maraging is a portmanteau of “martensite” and “aging”. The

aging heat treatment process results in the evolution of precipitates. PH 13-8 Mo consists of a ductile martensitic matrix, strengthened by the evolution of nanoscaled precipitates [6]. Unlike conventional PH and maraging steels where the strength is enhanced by the evolution of Cu-rich, Nb-rich, and/or  $\text{Ni}_3\text{Ti}$  particles [7], the strengthening precipitate in the PH 13-8 Mo steel is  $\beta$ -NiAl with an ordered CsCl structure [8,9]. NiAl precipitates are one of the most effective intermetallic phases to strengthen the maraging steels [1,10,11] due to their lattice constant of  $2.8752 \pm 0.0004 \text{ \AA}$  which is close to that of the ferritic steel ( $\alpha$ -Fe, bcc);  $2.8829 \pm 0.0002 \text{ \AA}$  [12]. As a result, a minimal lattice misfit between the precipitates and the matrix is achieved which will then enhances the strength through the chemical ordering effect and the creation of backstresses (the forces that oppose the dislocation motion) [13].

The importance of maraging steels in different industries makes it imperative to study their additive manufacturing (AM) since metal AM processes have been emerging in these industries as revolutionary

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manufacturing techniques [14]. The reason for identifying the metal AM processes as revolutionary manufacturing techniques is due to their capability to fabricate near-net-shape, complex-shaped, and on-demand components, directly from a feedstock (usually wire or powder). In most of the metal AM processes, the fabrication of the components is accomplished by melting the successive layers of feedstock material and joining them on top of each other [15]. The melting process is performed by employing a high-power energy source; in many cases either a laser beam or an electron beam [14,16]. Metal AM processes are classified into two main groups: directed energy deposition (DED) and powder bed fusion (PBF) [17]. Considering the most common energy sources (*i.e.* laser (L) and electron beam (EB)), the main AM processes for metallic materials can be categorized as L-PBF, EB-PBF, L-DED, and EB-DED [17]. Among these methods, L-PBF has shown a promising trend due to its capability to fabricate metallic components with enhanced properties. The enhanced properties are mainly achieved as a result of tailoring ultrafine, metastable, and hierarchical microstructures using the L-PBF process [18–21]. Such microstructures develop due to the very high solidification cooling rates (in the range of  $10^3$ – $10^8$  K/s) during the L-PBF process [15,22].

EOS GmbH has developed a low carbon Fe–Cr–Ni–Al maraging stainless steel powder in the family of PH 13-8 Mo [23] which can be used for the L-PBF fabrication. Some studies have been conducted on the L-PBF of CX steel in the as-built condition. Asgari and Mohammadi [24] fabricated L-PBF-CX components and studied the microstructure and mechanical properties in the as-built condition. They showed that dense L-PBF-CX parts with the least amount of defects could be fabricated and the yield strength (YS) and ultimate tensile strength (UTS) in the as-built condition can reach 1036 MPa and 1113 MPa, respectively. Saby et al. [25] reported the hardness of as-built L-PBF-CX as  $\sim 33$ HRC. Sanjari et al. [26], Shahriari et al. [27], and Pirgazi et al. [28] investigated the microstructure of the L-PBF-CX components with different geometries in the as-built condition. They reported a lath martensitic structure along with retained austenite, in the as-built condition. Both martensite lath size and retained austenite amount, and consequently, the mechanical properties changed by changing the geometry of the samples.

Since CX is a heat treatable steel, the feedstock vendor has suggested the following recipe as a generic treatment for precipitation hardening: austenitization at 900 °C for 1 h followed by rapid air cooling and aging at 530 °C for 3 h [23]. Some researchers have followed the exact procedure and others have conducted a different version of the heat treatment. Table 1 summarizes different heat treatments employed for the L-PBF-CX and the resulting strength and elongation to fracture. Elongation to fracture represents the material's ductility.

Referring to Table 1, strength enhancement after heat treatment is accompanied by ductility loss, which is a common phenomenon in the field of materials science and engineering and is known as the strength-

ductility trade-off. In addition, a range of strengths and ductilities has been reported for the heat-treated L-PBF-CX. Two reasons could have caused such variations; different heat treatment recipes and different starting microstructures. Various process parameters have been employed in the available studies to print the CX components with different geometries. It is well-established that changing the process parameters and sample geometry will affect the thermal boundaries during solidification which will then change the thermal history and the resulting microstructure [14,41]. In the case of L-PBF-CX, Sanjari et al. [26,42] showed that printing cuboid, vertical rod, and horizontal rod would substantially affect the microstructure in terms of lath martensite size, retained austenite amount, and texture. Hadadzadeh et al. [43] investigated the response of the cuboid L-PBF-CX to the standard austenitization-aging heat treatment. They observed that such a heat treatment would not result in the evolution of the  $\beta$ -NiAl precipitates and strength enhancement. In fact, the austenitization heat treatment substantially changed the microstructure (*i.e.* martensite lath growth and dislocation annihilation) and retarded the precipitation. On the other hand, direct aging of the as-built cuboid samples resulted in the evolution of nanometric  $\beta$ -NiAl precipitates and hardness enhancement. Meanwhile, the lath martensitic structure was refined due to the rearrangement of the dislocations and recovery, rather than dislocation annihilation. Considering these observations, it is obvious that the precipitation kinetics and the resulting strength are very sensitive to the starting microstructure. In addition, sophisticated microstructural changes occur in the L-PBF-CX during the aging heat treatment that can deteriorate the final strength. It is very important to preserve the dislocations during the heat treatment process since the dislocation networks in the L-PBF alloys can substantially increase both strength and ductility [21]. Therefore, it is essential to conduct a systematic study on the kinetics of precipitation formation and austenite reversion in L-PBF-CX. In a recent study, Shahriari et al. [44] conducted dilatometry on the L-PBF-CX at different heating rates to determine the transformation temperatures (*i.e.* precipitation and austenite reversion). However, they did not model the kinetics of these phase transformations.

In the current study, the kinetics of precipitation and austenite reversion phase transformations in an L-PBF-CX is studied using the differential scanning calorimetry (DSC) technique. DSC is a reliable and precise technique to determine the characteristics of phase transformations in terms of the reaction's enthalpy and transition temperatures [45]. Some studies have been conducted on the DSC of L-PBF maraging steels to determine the onset and end temperatures of precipitation reactions [46,47], however, none of them was dedicated to the modeling of the phase transformation kinetics. Since the microstructure of L-PBF alloys is different from the conventionally fabricated counterparts [14], their physical properties are also different [48]. Therefore, it is essential to consider the role of microstructure in the

**Table 1**

Summary of the heat treatment procedures employed for L-PBF-CX and the resulting strength and elongation to fracture (YS: yield strength, TS: tensile strength, El: elongation to fracture).

Heat Treatment		As-Built			Heat-Treated			Reference
Austenitization	Aging	YS (MPa)	TS (MPa)	El (%)	YS (MPa)	TS (MPa)	El (%)	
850 °C/30min	525 °C/4 h	–	–	–	1600	1720	1.7	[29]
		–	–	–	1560 <sup>a</sup>	1650 <sup>a</sup>	–	[30]
		–	–	–	1640 <sup>b</sup>	1700 <sup>b</sup>	–	–
		–	–	–	1600	1700	–	[31]
900 °C/1 h	530 °C/3 h	992	1058	18.4	1563	1683	8.4	[32]
900 °C/1 h	530 °C/3 h	694	1043	16.3	1528	1601	7.3	[33–37]
–	–	694	1043	–	1395	1510	9.3	–
850 °C/30min	525 °C/2 h	1006 <sup>a</sup>	1170 <sup>a</sup>	16.6 <sup>a</sup>	1533 <sup>a</sup>	1680 <sup>a</sup>	11.2 <sup>a</sup>	[38–40]
		919 <sup>b</sup>	1090 <sup>b</sup>	10.5 <sup>b</sup>	1600 <sup>b</sup>	1683 <sup>b</sup>	5.9 <sup>b</sup>	

<sup>a</sup> Horizontally-built.

<sup>b</sup> Vertically-built.

kinetics of phase transformation when a heat treatment is formulated. To the best of our knowledge, no kinetics modeling is available for the precipitation phase transformation in the L-PBF maraging steels. Likewise, the available heat treatment recipe(s) for L-PBF maraging steels seem to be developed by either adopting the recipes for the conventionally-fabricated counterparts or trial-and-error approaches. In both cases, the desired properties are not guaranteed. We believe the novelty of our work lies within our systematic approach to modeling the kinetics of phase transformation, using the kinetics modeling results to formulate the heat treatment, and heat treating the material to preserve the L-PBF microstructural features. Cylindrical rods of CX are fabricated using an EOS M290 machine. The samples are then characterized by employing the DSC technique to determine the onset and end temperatures of the precipitation and austenite reversion phase transformations. The kinetics of phase transformation is then modeled under non-isothermal and isothermal conditions using the modified Johnson-Mehl-Avrami-Kolmogorov (JMAK) model. The modeling results are then validated against experimental observations. The modeling results are also used to design a direct aging heat treatment to enhance the strength of the material at minimal ductility loss. These findings will pave the pathway toward optimum post-process heat treatments to obtain the desired strength-ductility balance in L-PBF precipitation-hardenable steels.

## 2. Experimental procedure

### 2.1. Materials and L-PBF process

Cylindrical rods of CX alloy were printed through the L-PBF technique using an EOS M290 machine. The nominal composition of the CX powder was Fe-12 wt% Cr-9.2 wt% Ni-1.6 wt% Al-1.4 wt% Mo-0.05 wt% C, with an average particle size of  $37.5 \pm 16 \mu\text{m}$  [24]. The rods were 12 mm in diameter and 120 mm in length, printed in a vertical configuration as shown schematically in Fig. 1. The process parameters were selected based on the optimum ones recommended by EOS GmbH to yield the least defects. The important process parameters included laser power ( $P$ ), scan speed ( $v$ ), hatch distance ( $l$ ), and layer thickness ( $h$ ). In the current study, they were set as  $P = 258.7 \text{ W}$ ,  $v = 1067 \text{ mm/s}$ ,  $l = 100 \mu\text{m}$ , and  $h = 30 \mu\text{m}$ . The laser spot size was  $100 \mu\text{m}$ , and the process was completed under an Ar atmosphere with a controlled oxygen content, held at  $<0.1\%$ . It should be noted that oxygen pickup during the L-PBF process is very minimal due to the protective argon atmosphere [49], therefore, in our analysis in this paper oxygen did not have a significant

role. The rods were then cut from the build plate using the wire EDM technique and used for the rest of the study in the as-built condition.

### 2.2. Differential scanning calorimetry (DSC) analysis

To study the phase transformations in the as-built L-PBF-CX rods, DSC tests were performed using SDT 650-TA Instruments (Delaware, USA). Small discs with the dimensions of 4 mm in diameter and 2 mm in height, were machined from the as-built rods, as shown in Fig. 1. These discs were then used for the DSC analyses. The samples were machined along the centerline of the rods, away from the bottom and top to ensure the consistency of the microstructure and chemistry. Since the microstructure and chemistry are consistent in horizontal rods printed using the EOS M290 machine [49,50], the DSC results were consistent and repeatable. Each DSC trial was repeated 5 times and for each trial, a new as-built sample was used. Each DSC curve was evaluated and the characteristics (i.e., onset, peak, and end temperatures) were reported as an average of the 5 trials. The DSC curves reported in the paper are representative ones. The DSC runs were performed under various heating rates of  $\Phi = 5, 10, 20, 30, 40$ , and  $50 \text{ }^\circ\text{C/min}$  to a maximum temperature of  $1000 \text{ }^\circ\text{C}$ . The DSC curves of 5 and  $10 \text{ }^\circ\text{C/min}$  heating rates were excluded from further analyses since the peaks were weak and it was not possible to determine the onset and end temperatures accurately. Such a challenge is a common issue with low heating rates and has been reported previously [51]. All the DSC tests were performed under 99.999% argon which flowed throughout the run at  $100 \text{ mL/min}$ . The DSC curves were then analyzed to identify the exothermic and endothermic reactions that occur in the material.

### 2.3. Mechanical and microstructural characterization

A Wilson Rockwell hardness tester with a diamond indenter under 150 kgf load and a dwell time of 5s were used to measure the hardness of the as-built and heat-treated specimens. Tensile testings were conducted following the ASTM E8/E8M-21. Dogbone samples with a gauge length of 15.5 mm and a diameter of 6 mm were machined from the rods and tested using an Instron 8800 Uniaxial Servohydraulic System. The tensile tests were conducted under the displacement-controlled condition using an MTS extensometer with a speed of  $0.05 \text{ mm/min}$ .

Thin films for TEM were prepared using the focused ion beam (FIB) lift-out approach by employing a Thermo Scientific Helios 5 UC Dual-Beam, as shown in Fig. 2. With an X-FEG source and a high tension adjustable from 80 to 200 kV, TEM studies were carried out using the Thermo Scientific Talos 200X. This system was configured with four in-column SDD Super-X detectors for energy-dispersive X-ray spectroscopy (EDS) signal detection and compositional mapping of the samples.

## 3. Results

### 3.1. DSC curves

Fig. 3 shows the representative DSC curves of L-PBF-CX at different heating rates. The DSC curves are reported in terms of normalized heat flow ( $\text{W/g}$ ) with respect to both temperature and time. It is obvious that by increasing the heating rate, the process is completed over a shorter time. The DSC curves are featured by various endothermic and exothermic peaks, indicating the solid-state phenomena/phase transformations in the material. In the current study, two major peaks are considered for further analyses; the exothermic peak over the temperature range of  $\sim 410^\circ\text{C}$ – $610^\circ\text{C}$  and the endothermic peak over the temperature range of  $\sim 615^\circ\text{C}$ – $760^\circ\text{C}$ . The former represents the precipitation and the latter represents austenite reversion [44]. The reason for considering these two phenomena is the fact that while precipitation significantly enhances the strength of the L-PBF-CX, austenite reversion negatively affects the strength [43]. Since the onset temperature of austenite reversion is very close to the end temperature of precipitation,

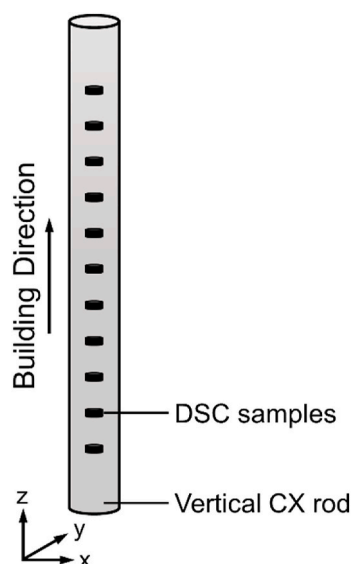


Fig. 1. Schematic of a vertical L-PBF-CX rod along with the DSC samples.

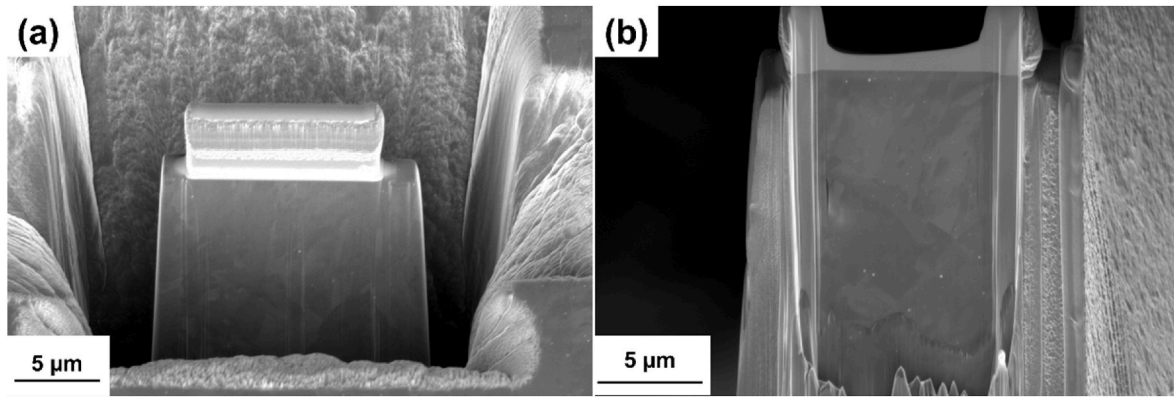


Fig. 2. (a) ion-milled trenches around the FIB sample and (b) thinned TEM lamella.

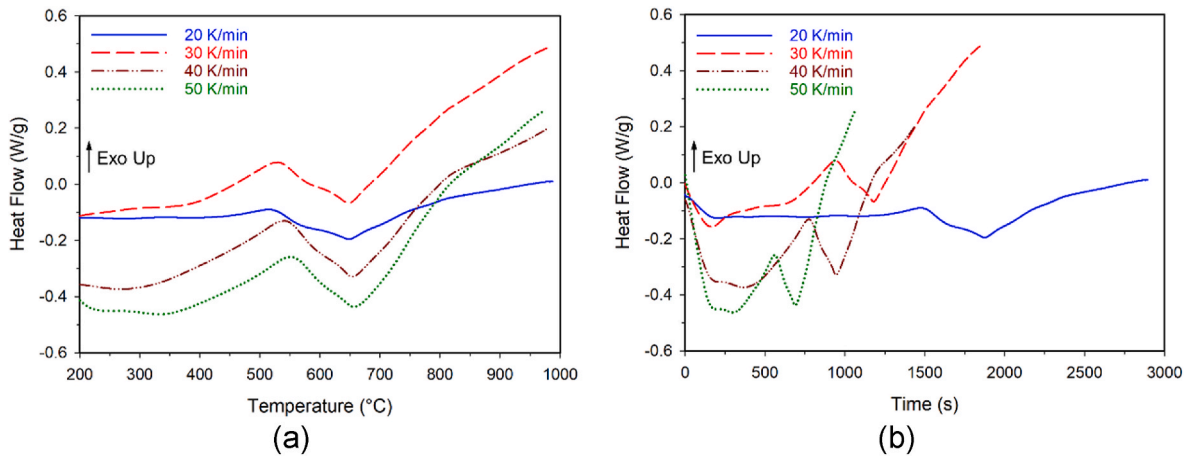


Fig. 3. DSC curves of L-PBF-CX with respect to (a) temperature and (b) time.

reverted austenite may form during the aging heat treatment. Therefore, it is essential to consider the kinetics of austenite reversion to count for it during designing the aging heat treatment for L-PBF-CX.

The analysis of the endothermic and exothermic peaks in the DSC curves requires the subtraction of the baseline [51]. The baseline represents the part of the curve that does not exhibit any transitions or reactions. Therefore, it should be subtracted from the DSC curve to obtain the actual heat flow associated with the phase transformation. The first step of the baseline subtraction is to determine the onset temperature ( $T_o$ ) and end temperature ( $T_e$ ) of the reaction. Since the baseline is temperature and time-dependent [52], the determination of  $T_o$  and  $T_e$  should be conducted for each heating rate separately. Once these characteristics are determined, the baseline is constructed considering the fact that  $T_o$  and  $T_e$  represents the start and end points of the baseline and then it is subtracted from the actual DSC curve [53].  $T_o$  and  $T_e$  can be determined by taking the first derivative of the heat flow with respect to the temperature and identifying the points where the slope is changing abruptly, as shown in Fig. 4. After identifying  $T_o$  and  $T_e$ , the peak analyze module in the OriginPro 2021 software was employed for baseline subtraction [48]. Fig. 5 shows the processed peaks (after baseline subtraction) for the precipitation and austenite reversion phase transformations in the L-PBF-CX.

An important characteristic of the exothermic and endothermic peaks in any DSC analysis is the peak temperature, where the transformation reaches the maximum rate. Fig. 6 shows the effect of the heating rate on the peak temperatures of both transformations. By increasing the heating rate the peak temperature increases which is a common phenomenon in DSC studies [51]. The peak temperatures for precipitation and austenite reversion vary between 515–554 °C and

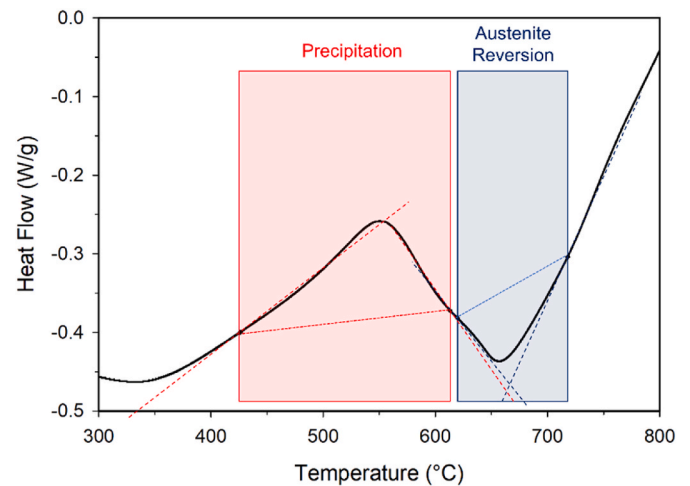


Fig. 4. Determination of onset and end temperatures of precipitation and austenite reversion peaks.

647–657 °C, respectively.

### 3.2. Degree of transformation and activation energy

The degree of phase transformation (transformed fraction,  $f$ ) at any time/temperature can be determined by using the processed DSC curves (Fig. 5). The transformed fraction is equivalent to the fraction of heat



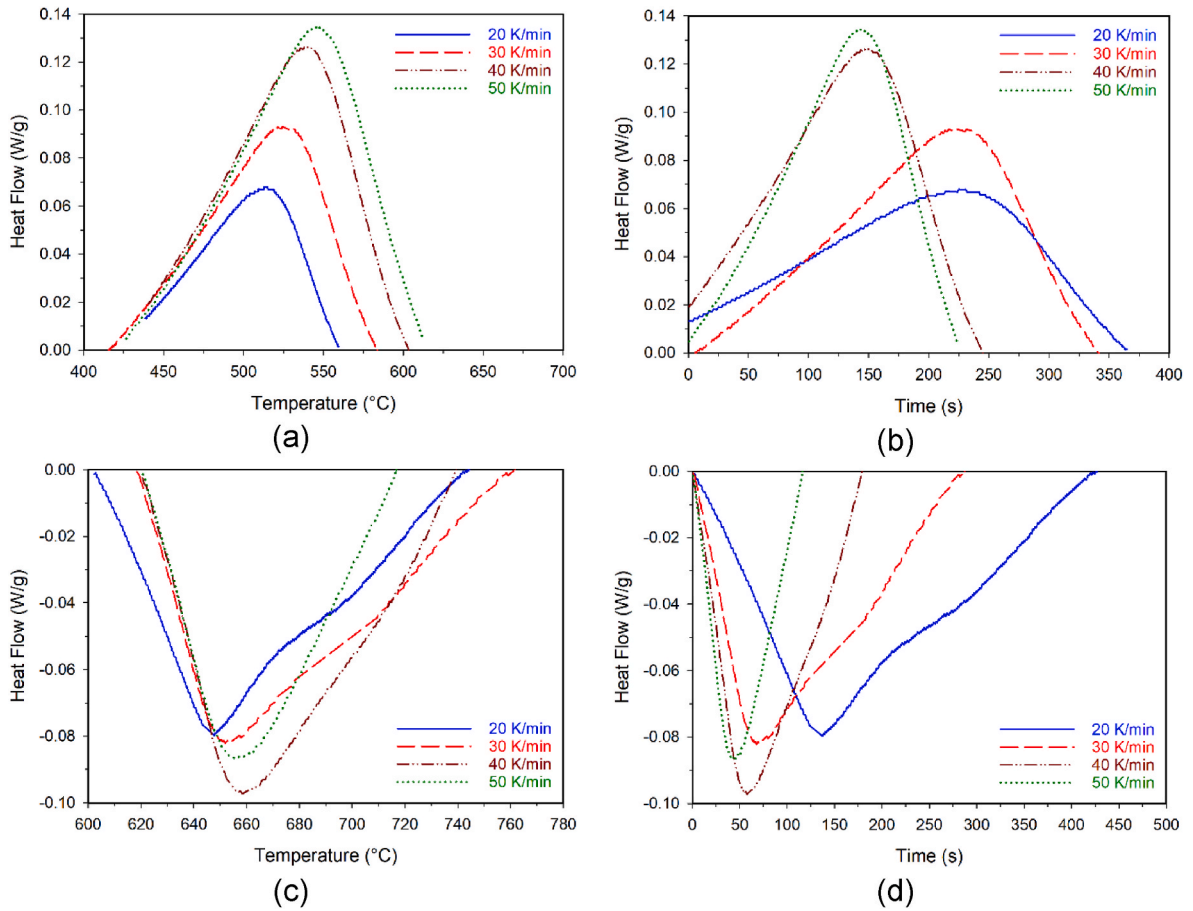


Fig. 5. Processed DSC peaks for precipitation with respect to (a) temperature and (b) time, and austenite reversion with respect to (c) temperature and (d) time.

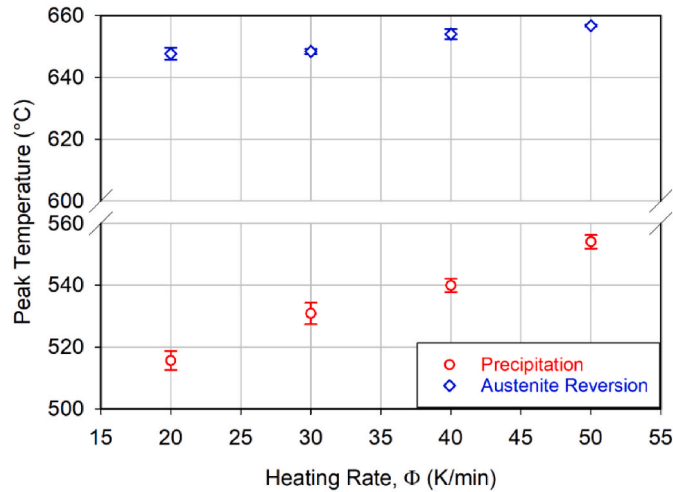


Fig. 6. The effect of heating rate on the peak temperatures of precipitation and austenite reversion.

that is absorbed/released during the transformation process and can be determined as follows [51],

$$f(t) = \frac{\int_{t_0}^t H dt}{\int_{t_0}^{t_e} H dt} \quad (1)$$

where  $f(t)$  is the transformed fraction at any given time ( $t$ ),  $H$  is the heat flow measured during the DSC, and  $t_0$  and  $t_e$  are the transformation onset

and end times, respectively, associated with the onset and end temperatures. The denominator in Eq. (1) represents the total enthalpy of the transformation equivalent to the area under the processed peak curves in Fig. 5 (b) and (d). The results for both precipitation and austenite reversion transformations are plotted in Fig. 7. In both cases, the transformation follows the common sigmoidal trend. Moreover, by increasing the heating rate, the phase transformation is completed over a shorter period [51].

Determination of the kinetics model parameters begins with the

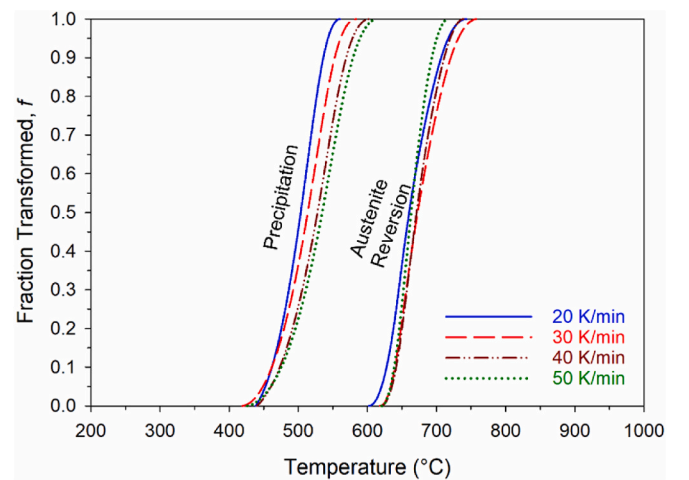


Fig. 7. Calculated degree of precipitation and austenite reversion using the DSC results at different heating rates.

evaluation of the transformation activation energy ( $E_a$ ).  $E_a$  is a thermodynamics parameter that represents the minimum energy required for the initiation of the process. In the case of the availability of non-isothermal DSC results at different heating rates,  $E_a$  can be determined using the modified Kissinger method [48,51,54], as follows,

$$\ln \frac{T_p^2}{\Phi} = \frac{E_a}{RT_p} + \text{constant} \quad (2)$$

where  $T_p$  is the process peak temperature (in K) (Fig. 6),  $\Phi$  is the heating rate (in K/s),  $R$  is the universal gas constant (8.314 J/mol.K), and  $E_a$  is the activation energy in J/mol. Referring to Eq. (2), the activation energy can be determined by linear regression between  $\ln \frac{T_p^2}{\Phi}$  and  $\frac{1000}{T_p}$  data as shown in Fig. 8. The slope of the line is equivalent to  $-\frac{E_a}{R}$ , where the activation energy can be then determined. In the current study, the activation energy for precipitation and austenite reversion was evaluated as 118.6 kJ/mol and 591.1 kJ/mol, respectively. The activation energy for precipitation in a conventionally manufactured counterpart (i.e. investment cast PH 13-8 Mo stainless steel) was evaluated as 139 kJ/mol [55] which is higher than that of the L-PBF-CX. The main reason for such a difference is the higher dislocation density in the L-PBF alloys [21,43,56] compared to the conventionally manufactured ones. The high-density dislocations act as pipelines for the diffusion of the alloying elements [57] and facilitate the nucleation and growth processes [58]. As a result, the precipitation activation energy is lowered.

### 3.3. Non-isothermal and isothermal phase transformation kinetics parameters

The kinetics of phase transformations under non-isothermal heating conditions is usually studied by employing the modified Johnson-Mehl-Avrami-Kolmogorov (JMAK) model [51]. The common JMAK model has been widely used to investigate the isothermal phase transformations [59–62], as follows

$$f = 1 - \exp [-(kt)^n] \quad (3)$$

where  $f$  is the fraction transformed at time  $t$ ,  $n$  is the Avrami constant, and  $k$  is the reaction rate. For the isothermal processes,  $k$  is independent of the temperature. However, in the modified JMAK model, a temperature-dependent reaction rate has been established as follows [63,64]

$$k(T) = k_0 \exp \left( -\frac{E_a}{RT} \right) \quad (4)$$

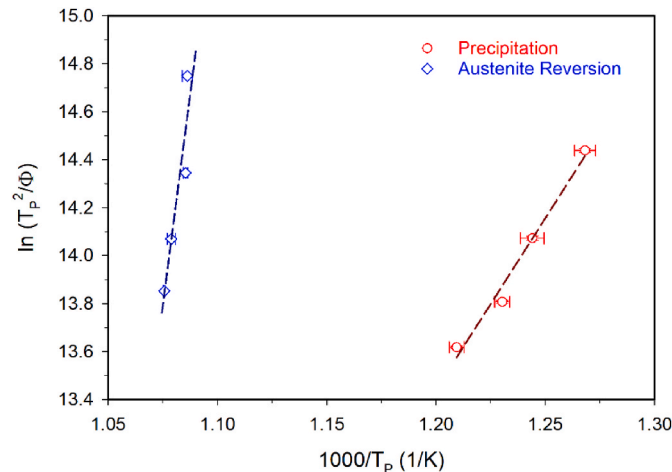


Fig. 8. Kissinger plot for precipitation and austenite reversion to determine the activation energy.

where  $k(T)$  is the temperature-dependent reaction rate and  $k_0$  is the pre-exponential factor.  $E_a$  should be determined using the Kissinger method (Eq. (2) and Fig. 8). Therefore, the general JMAK model can be derived as follows

$$f = 1 - \exp \left[ - \left( \int_{t_0}^t k(T) dt \right)^n \right] \quad (5)$$

The general JMAK equation (Eq. (5)) is valid for a constant temperature, however, for continuous heating with a heating rate of  $\Phi = \frac{dT}{dt}$ , the JMAK equation is revised as follows [63,65,66]

$$f = 1 - \exp \left[ - \left( \left( \frac{1}{\Phi} \right) \int_{T_0}^T k(T) dT \right)^n \right] \quad (6)$$

By implementing  $k(T)$  from Eq. (4) into Eq. (6), the modified JMAK equation to model the non-isothermal phase transformation kinetics at any temperature  $T$  (in K) is derived as follows [67,68]

$$f = 1 - \exp \left[ - \left( \frac{k_0}{\Phi} \left( \exp \frac{-E_a}{RT} \right) (T - T_0) \right)^n \right] \quad (7)$$

where  $\Phi$  is the heating rate (in K/s or °C/s) and  $T_0$  is the onset temperature of the phase transformation determined from the DSC curves. The values for  $k_0$  and  $n$  were determined by curve fitting to the experimental transformed fraction plots (Fig. 7) obtained from the DSC for each heating rate, by employing the activation energy values evaluated using the Kissinger method. In the next step, an average was taken over the  $k_0$  and  $n$  values to determine the corresponding parameters for the precipitation and austenite reversion processes [51]. The parameters were determined as  $n = 0.95$  and  $k_0 = 2.85 \times 10^5$  for precipitation and  $n = 0.40$  and  $k_0 = 3.28 \times 10^{30}$  for austenite reversion. These values can be then employed to model the kinetics of precipitation and austenite reversion during an isothermal aging process by using Eq. (3) [51]. To determine the value for the reaction rate ( $k$ ) at the aging temperature, Eq. (4) could be employed.

### 3.4. Design of heat treatment

The results of the isothermal kinetics models were employed to age the as-built L-PBF-CX at different temperatures in the range of 400–600 °C to assess the model's efficiency. The hold time at each temperature was determined by employing the isothermal kinetics modeling results to achieve 100% precipitate fraction transformation. The aging conditions are summarized in Table 2.

The hardness of the samples was measured after the aging treatments and the results are shown in Fig. 9. As observed, the hardness of the as-built-CX was  $34.9 \pm 2.9$  HRC, which is in agreement with those reported previously [32,69]. After aging the material at 400–600 °C, the hardness increased due to the formation of  $\beta$ -NiAl precipitates. The extreme aging temperatures, i.e. 400 °C and 600 °C did not result in the maximum hardness of the material. It appears that the precipitates that formed at 400 °C did not possess the characteristics of effective precipitates such as size and coherency, which requires further microstructural analyses. On the other hand, at 600 °C precipitate coarsening may have occurred [1] which resulted in hardness reduction. In addition, the formation of extensive reverted austenite at 600 °C (close to 17%, based on the kinetics models) could be another reason for hardness deterioration. The aging heat treatments at 460–550 °C resulted in hardness values in the range of 50–51 HRC, which is the common hardness of hardened L-PBF-CX [30–32,69]. Therefore, the isothermal kinetics modeling

Table 2  
Aging conditions of L-PBF-CX.

Aging temperature (°C)	400	460	495	530	550	600
Aging time (min)	1667	1078	424	180	115	40

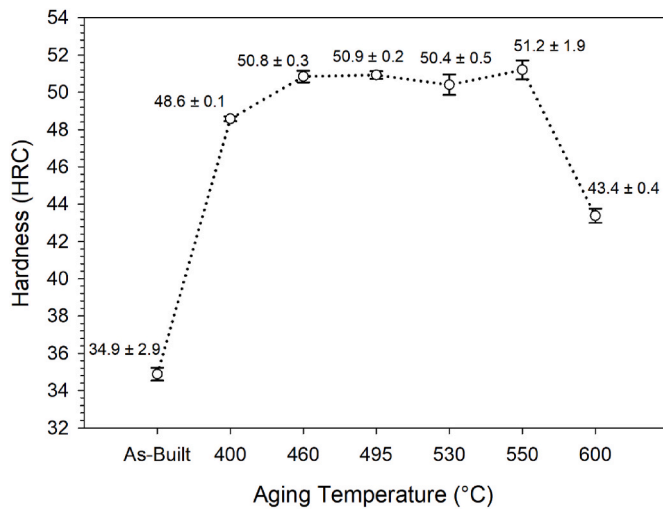


Fig. 9. Hardness of L-PBF-CX after aging heat treatment.

results could be successfully employed to determine the aging temperature and time for the L-PBF-CX. In addition, the L-PBF-CX could be aged over a range of temperatures (460–550 °C) rather than a single temperature (530 °C). Referring to the hardness results, it appears that the optimum aging condition is 550 °C/115 min rather than the common 530 °C/180 min, which resulted in the maximum hardness for the range of studied conditions.

### 3.5. Strength, ductility, and microstructure of heat-treated L-PBF-CX

By employing the optimum aging condition (i.e. 550 °C/115 min), the L-PBF-CX rods were directly aged. Fig. 10 shows the tensile properties of the as-built and heat-treated (directly aged) L-PBF-CX. As seen, the as-built material possesses yield and tensile strengths of 929 MPa and 1032 MPa, respectively with a total elongation of 15.3%. These properties are consistent with those reported previously by other researchers as listed in Table 1. After direct aging at 550 °C for 115 min, the yield and tensile strengths significantly improved to 1659 MPa and 1738 MPa, while the total elongation to fracture slightly reduced to 14.7%. Both strength enhancement and ductility retention are phenomenal for the heat-treated L-PBF-CX, as shown in Fig. 11. Referring to Fig. 11 (a), the strength of heat-treated L-PBF-CX in this study is the highest among the available values in the literature. On the other hand, while the heat treatment of L-PBF-CX generally results in a

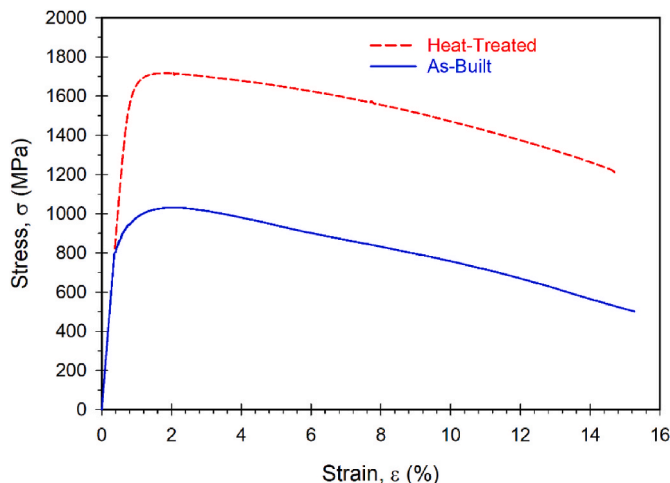


Fig. 10. Tensile properties of as-built and heat-treated L-PBF-CX.

reduction in ductility, as seen in Fig. 11 (b), the suggested heat treatment in our study did not deteriorate the ductility. In other words, the accurate aging of the L-PBF-CX maraging steel considering the kinetics of phase transformations resulted in the strength-ductility trade-off enhancement. It appears from the results that preserving the original microstructure developed during the L-PBF process is the governing mechanism in strength enhancement and ductility retention.

To further analyze the effect of direct aging, the microstructure of the heat-treated material was studied using TEM. Fig. 12 shows the TEM bright field (BF) of the heat-treated L-PBF-CX with details of the lath martensitic structure and the dislocation networks. The martensite laths are featured by the networks of entangled dislocations. These are the pre-existing dislocation networks, developed during the rapid solidification in the L-PBF process, and preserved during the low-temperature aging treatment. Apparently, this TEM micrograph does not reveal any evidence of  $\beta$ -NiAl formation. Due to the ultrafine size of  $\beta$ -NiAl precipitates which is  $< 10$  nm [11] and the full coherency between these precipitates and the martensitic matrix [43], the lattice misfit strain is  $< 1\%$  [12]. Therefore, revealing them using TEM imaging is challenging. However, superlattice reflections are observed in the selected area diffraction (SAD) patterns due to the ordered structure of the  $\beta$ -NiAl precipitates [11,43].

With the beam aligned parallel to the  $[001]_{Fe/\beta-NiAl}$  direction, an SAD pattern was obtained from one of the martensite laths as shown in Fig. 13 (a) and (b). The main diffraction spots associated with the martensitic matrix along with the weak superlattice spots associated with the  $\beta$ -NiAl precipitates are visible in the SAD pattern. The corresponding dark-field (DF) image taken from a superlattice reflection is shown in Fig. 13 (c). The DF image reveals the nano-scaled  $\beta$ -NiAl precipitates dispersed in the lath martensite. The evolution of the nanometric precipitates is the main reason for strength enhancement in the aged L-PBF-CX [43].

The microstructure of the heat-treated L-PBF-CX was also analyzed in the scanning mode of TEM (STEM), and the STEM-BF image along with the corresponding EDS (energy dispersive X-Ray spectroscopy) maps of Al and Ni are shown in Fig. 14. The STEM-BF image of the material does not exhibit any evidence of the existence of the  $\beta$ -NiAl precipitates, as observed in the TEM images as well. However, the EDS elemental maps exhibit a good distribution of the nanometric precipitates in the matrix. Referring to the STEM-BF image, more details of the pre-existing dislocation networks are observed. These networks are interconnected and distributed in the microstructure of the material. These dislocations evolved as a result of the rapid solidification of the melt pools during the L-PBF process [43]. The presence of the pre-existing dislocation networks has a crucial effect on the mechanical properties of L-PBF steels since they can effectively improve ductility [70] by increasing the dislocation storage capability [71]. In addition, the dislocation networks can effectively facilitate precipitation since they act as pipelines for the diffusion of the substitutional elements [57]. Therefore, preserving the dislocations during any post-processing heat treatment of L-PBF steels is critical to ensure their benefits on the physical and mechanical properties. The employment of a low-temperature aging treatment in the current study by avoiding the high-temperature austenitization step effectively preserved the dislocation networks that resulted in the enhancement of both strength and ductility.

## 4. Discussion

The low-temperature aging heat treatment of the L-PBF-CX had two crucial roles in breaking the strength-ductility trade-off; preserving the dislocation networks and evolution of the nano-scaled  $\beta$ -NiAl precipitates. The microstructure of L-PBF-CX in the as-built condition with the same geometry and process parameters has been reported by Sanjari et al. [42]. The TEM analysis of the as-built L-PBF-CX revealed the entangled pre-existing dislocation networks in the microstructure of the material. Meanwhile, no evidence of  $\beta$ -NiAl precipitates was observed.



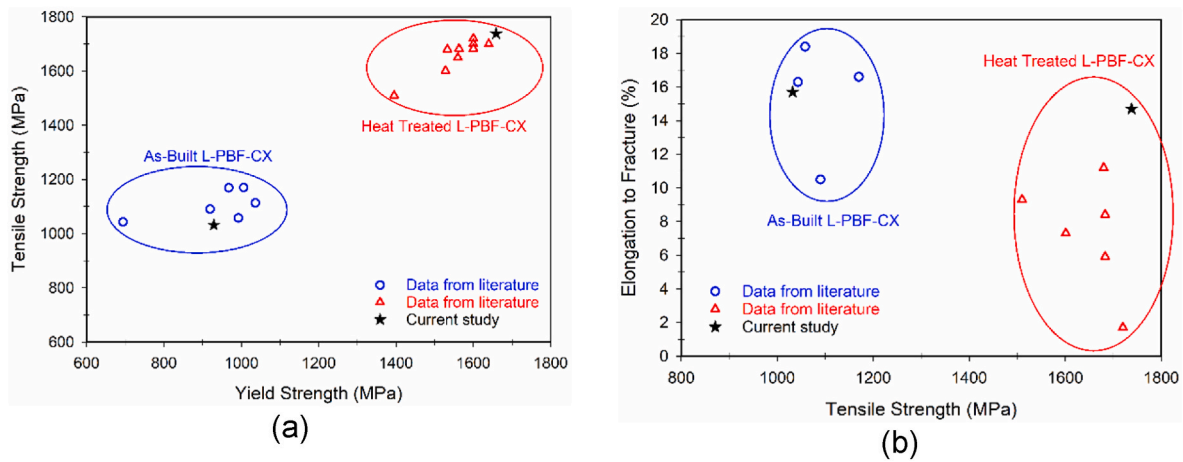


Fig. 11. Summary of strength and ductility of as-built and heat-treated L-PBF-CX, data from the literature are summarized in Table 1.

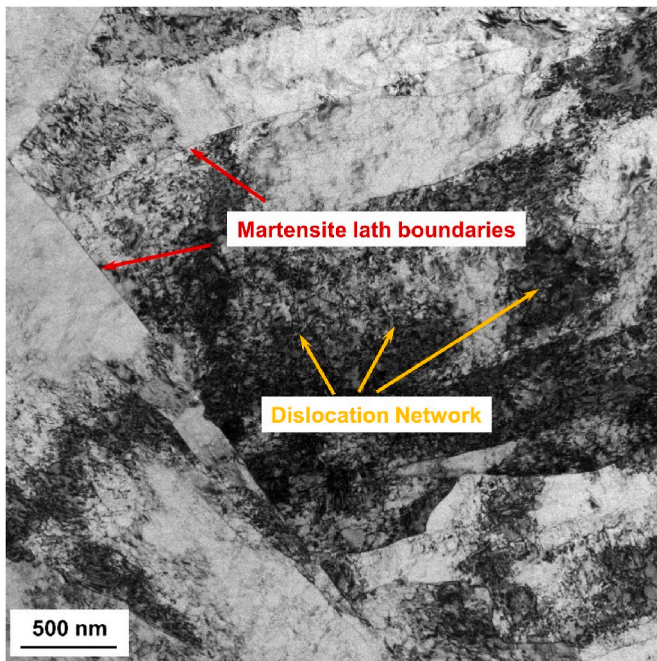


Fig. 12. TEM micrograph of heat-treated L-PBF-CX, showing the details of lath martensitic structure and the network of entangled dislocations.

The absence of the hardening precipitates in the as-fabricated (non-heat treated) maraging stainless steels is a common phenomenon reported by other researchers as well [11]. Therefore, the main strengthening mechanisms in the as-built L-PBF-CX are sub-boundary hardening (controlled by the martensite laths) and dislocation hardening (controlled by the pre-existing dislocation networks).

The TEM results reported by Sanjari et al. [42] for the as-built material and those reported in the current study for the heat-treated sample (Fig. 14) were employed to calculate the dislocation density. It should be noted that samples by Sanjari et al. [42] were fabricated using the same process parameters and geometry on the same printing machine as the current study. The dislocation density was determined by the method proposed by Pesicka et al. [72] and Rojas et al. [73]. The square grid lines were superimposed on the STEM micrographs and the number of horizontal ( $n_h$ ) and vertical ( $n_v$ ) dislocations-grid lines intersections were counted. The thickness of the TEM foil ( $t$ ) was determined using electron-energy loss spectroscopy (EELS) log ratio technique [74], and the dislocation density was calculated as follows,

$$\rho_d = \frac{1}{t} \left( \frac{\sum n_v}{\sum L_v} + \frac{\sum n_h}{\sum L_h} \right) \quad (8)$$

where  $\rho_d$  is the dislocation density (in  $\text{m}^{-2}$ ) and  $\sum L_h$  and  $\sum L_v$  are the total length of the horizontal and vertical lines, respectively. The values of the parameters used in Eq. (8) are shown in Table 3.

Using this method, the dislocation density in the as-built and heat-treated L-PBF-CX was estimated as  $1.2 \times 10^{15} \text{ m}^{-2}$  and  $1.08 \times 10^{15} \text{ m}^{-2}$ , respectively. As observed, the low-temperature aging preserved 90% of the pre-existing dislocations. Therefore, these dislocations can

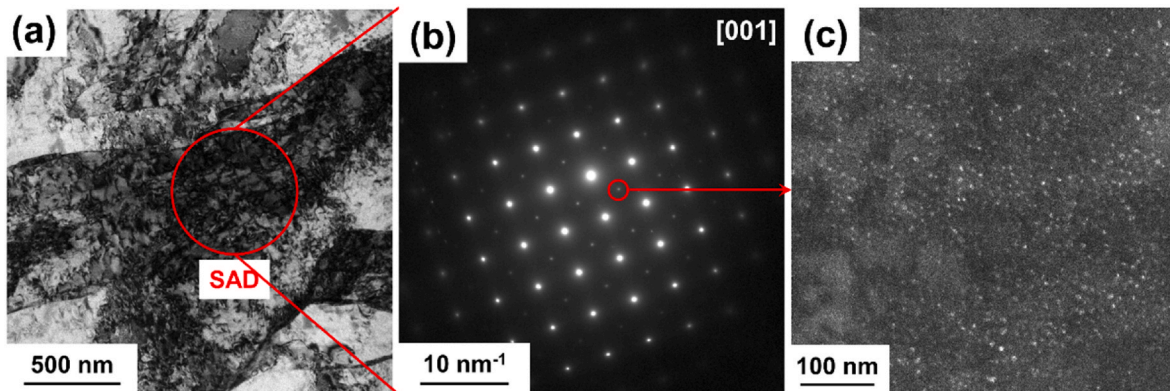


Fig. 13. (a) TEM-BF image of heat-treated L-PBF-CX, (b) SAD showing the main diffraction spots from the martensitic matrix and superlattice spots from  $\beta$ -NiAl nanoscaled precipitates, and (c) dark-field image of the matrix using the  $\beta$ -NiAl superlattice reflection.



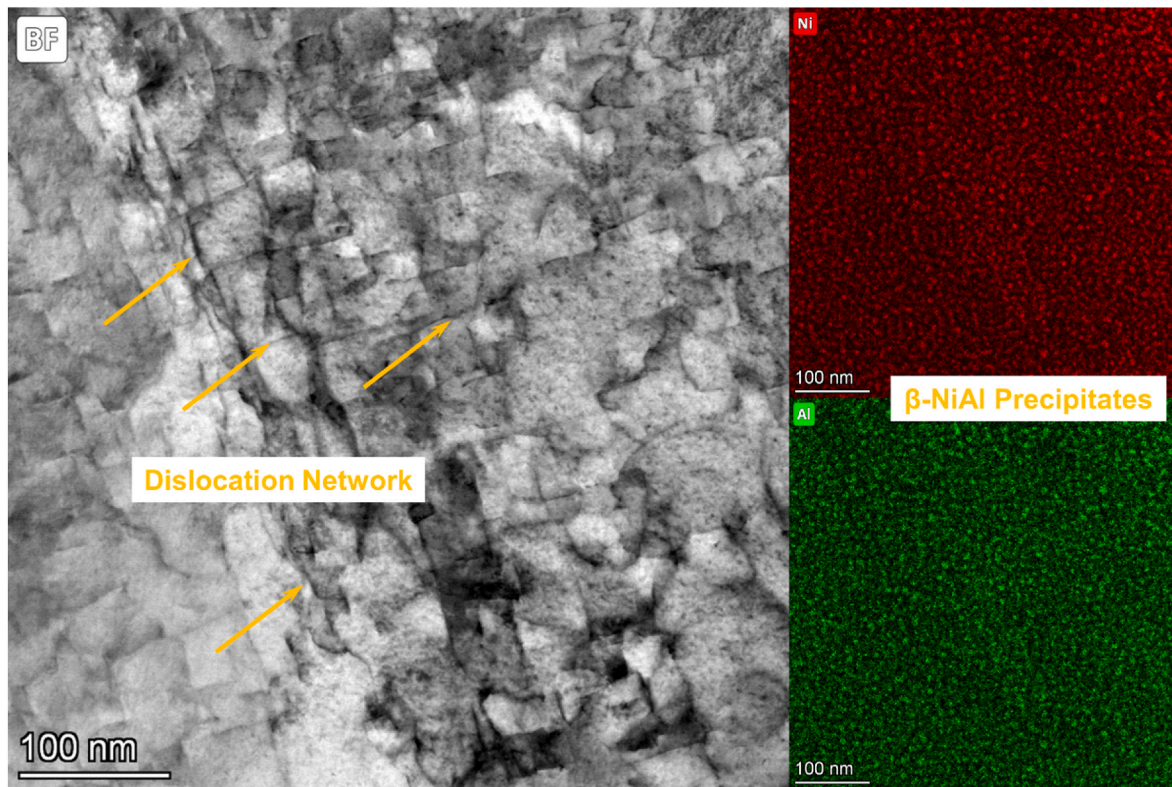


Fig. 14. Left: STEM-BF image of the heat-treated L-PBF-CX, right: corresponding EDS elemental maps of Ni and Al.

**Table 3**  
Quantities used in Eq. (8).

	$\Sigma n_v$	$\Sigma L_v$ (nm)	$\Sigma n_h$	$\Sigma L_h$ (nm)	$t$ (nm)	$\rho_d$ (m <sup>-2</sup> )
As-built	39	1727	65	1727	50	$1.2 \times 10^{15}$
Heat-treated	34	804	27	804	70	$1.08 \times 10^{15}$

effectively contribute to both strength and ductility. The survey of the mechanical properties obtained in the current study and those available in the literature (Fig. 11) reveals a significant difference in the ductility of the heat-treated L-PBF-CX. We believe that the preserved dislocations are the main reason for ductility retention. Liu et al. [70] showed that the pre-existing dislocation networks in the L-PBF steels maintain their configuration during the plastic deformation and act as an ideal “modulator” to slow down the dislocation motion. The in-situ TEM study by Liu et al. [70] showed that these networks significantly, and not fully, impede the dislocation motion during the plastic deformation. Therefore, these networks increase the dislocation storage capacity of the material which results in ductile plastic deformation. The preserved pre-existing dislocation networks in the current study are playing the same role; they maintain the dislocation storage capacity of aged L-PBF-CX during the plastic deformation and retain ductility.

Using the TEM-EDS elemental maps (Fig. 14) the average size, volume fraction, and inter-particle distance of  $\beta$ -NiAl precipitates were estimated as 8.8 nm, 3.4%, and 13.9 nm, respectively. The strength increment due to the evolution of nano-scaled  $\beta$ -NiAl precipitates ( $\Delta\sigma_p$ ) is governed by three factors [43,75]: order strengthening ( $\Delta\sigma_o$ ), coherency strengthening ( $\Delta\sigma_c$ ), and modulus mismatch strengthening ( $\Delta\sigma_m$ ), as follows,

$$\Delta\sigma_p = \Delta\sigma_o + \Delta\sigma_c + \Delta\sigma_m = 0.4M \frac{\gamma_{APB}}{b} \left( \frac{3\pi f}{8} \right)^{1/2} + 2.6M(\varepsilon G)^{3/2} \left( \frac{2rf}{Gb} \right)^{1/2} + 0.0055M(\Delta G)^{3/2} \left( \frac{2f}{Gb^2} \right)^{1/2} b \left( \frac{r}{b} \right)^{\frac{3m}{2}-1} \quad (9)$$

where  $M$  is the Taylor factor,  $\gamma_{APB}$  is the anti-phase boundary (APB) energy of the  $\beta$ -NiAl,  $b$  is the Burgers vector,  $f$  is the volume fraction of  $\beta$ -NiAl,  $G$  is the shear modulus of the matrix,  $r$  is the mean radius of  $\beta$ -NiAl,  $\varepsilon$  is the lattice mismatch parameter,  $\Delta G$  is the shear modulus mismatch between  $\beta$ -NiAl and matrix, and  $m$  is a constant. Table 4 summarizes the parameters used in determining the  $\beta$ -NiAl precipitation strength increment. The modeled strength increment in the aged L-PBF-CX due to the evolution of nano-scaled  $\beta$ -NiAl precipitates was estimated as  $\Delta\sigma_p = 779$  MPa. Referring to the tensile properties reported in Fig. 10 and Fig. 11, the experimental strength increment ( $\Delta\sigma_y$ ) after heat treatment was 730 MPa. The marginal discrepancy between the modeled and experimental values stems from two main reasons: (1) strength reduction due to the annihilation of 10% of the dislocations during the heat treatment which compensates for a portion of such a difference and (2) the nature of a modeling approach where some simplifying assumptions are made. Nevertheless, the precipitates are playing a significant role in strength enhancement in the heat-treated L-PBF-CX.

**Table 4**  
Parameters used in Eq. (9).

Parameter	Unit	Value	Reference
$M$	–	2.9	[11]
$\gamma_{APB}$	mJ m <sup>-2</sup>	500	[11]
$b$	nm	0.25	[43]
$f$	–	0.034	Current study
$\varepsilon$	–	0.0064	[43]
$G$	GPa	77	[11]
$r$	nm	4.4	Current study
$\Delta G$	GPa	4	[11]
$m$	–	0.85	[11]

## 5. Conclusions

In the current study, differential scanning calorimetry (DSC) was employed to determine the heat flow curves in a laser-powder bed fused (L-PBF) CX maraging steel (belonging to the PH 13-8 Mo family). The DSC runs were conducted at different heating rates of 20–50 °C/min to a maximum temperature of 1000 °C. The DSC curves were then analyzed to identify the peaks associated with precipitation and austenite reversion phase transformations. The former process exhibited an exothermic peak while the latter exhibited an endothermic peak. The peaks were processed to subtract the baseline and in the next step the onset, peak, and end temperatures of each phase transformation as a function of heating rate were determined. The heat flow associated with each processed peak was used to determine the degree of phase transformation during precipitation and austenite reversion. The peak temperatures were used to determine the activation energy of the phase transformation using the Kissinger method. The activation energy for precipitation and austenite reversion was evaluated as 118.6 kJ/mol and 591.1 kJ/mol, respectively. The activation energy for precipitation in L-PBF-CX was lower than that of a conventionally manufactured counterpart (139 kJ/mol). The main reason for such a difference is the higher dislocation density in the L-PBF-CX. In the next step, the modified Johnson-Mehl-Avrami-Kolmogorov (JMAK) model was employed to model the kinetics of phase transformations during non-isothermal heating. The good agreement between the experimental and modeled degree of transformation confirmed the suitability and effectiveness of the JMAK model. Therefore, this model was employed to predict the degree of precipitation and austenite reversion during isothermal aging at 400–550 °C. The material was aged over this range of temperatures and the hardness was measured. It was observed that the optimum aging condition is 550 °C/115 min rather than the common 530 °C/180 min, which resulted in the maximum hardness for the range of studied conditions. The tensile properties of the heat-treated L-PBF-CX at 550 °C/115 min were evaluated. It was observed that direct aging of L-PBF-CX enhanced the yield strength from 929 MPa to 1659 MPa and the tensile strength from 1032 MPa to 1738 MPa. Meanwhile, the elongation to fracture slightly decreased from 15.3% in the as-built condition to 14.7% after heat treatment. Such a phenomenal achievement in breaking the strength-ductility trade-off in the L-PBF-CX maraging steel was due to the evolution of nanometric  $\beta$ -NiAl precipitates while the entangled dislocation networks were preserved during the low-temperature heat treatment. The precipitates enhanced the strength and the dislocations retained the ductility.

## CRedit authorship contribution statement

**Richard Fabian:** Methodology, Validation, Investigation, Writing – original draft. **Amir Hadadzadeh:** Conceptualization, Methodology, Validation, Investigation, Resources, Writing – review & editing, Supervision.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Data availability

The data that has been used is confidential.

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